# **Carbohydrates Identification Lab**

Name

Date

Block \_\_\_\_

Purpose: To identify carbohydrates using Lugol's and Benedict's chemical tests.

**Background:** Scientists use a combination of biology and chemistry for their understanding of life and life processes The foods you eat are made of organic compounds. You can perform chemical tests to learn what foods contain carbohydrates, lipids, and/or proteins. Carbohydrates are an energy source or are used in making cell structures. The first tests you will be doing will be designed to specifically test for carbohydrates a.k.a sugars. These are also called <u>saccharides</u>. Carbohydrates are often classified according to the number of saccharide units they contain. Carbohydrates are molecules that contain the atoms **Carbon** (C), **Hydrogen** (H) and **Oxygen** (O) at a 1:2:1 ratio. (Example: glucose-  $C_6H_{12}O_6$ , table sugar –  $C_{12}H_{24}O_{12}$ )

- Saccharide is a term derived from the Latin for sugar (origin = "sweet sand")
- A monosaccharide contains a single carbohydrate
- A disaccharide contains two carbohydrate units
- A **polysaccharide** contains many carbohydrates, examples are starch and cellulose.



Benedict's Reagent	a positive reaction indicates the presence of a <b>reducing sugar (monosaccharide)</b> the precipitate may vary in color from green to orange, but <b>a strong reaction is</b> <b>brick red</b> .
Iodine	a positive reaction indicates the presence of a <b>starch</b> (polysaccharide) - a strong reaction is an <b>blue-black color</b>

Pre-lab Questions: Use the background information and your notes to help you answer the following question:

1. What are the three categories of carbohydrates discussed in your text-book and notes?

2. What does "mono" mean?\_\_\_\_\_

3. What does "di" mean? \_\_\_\_\_\_

- 4. What does "poly" mean? \_\_\_\_\_
- 5. What does "saccharide" mean?
- 6. Why are the prefixes "mono", "di", and "poly" and the base "saccharide" good to use when describing the three types of carbohydrates?

7. What are the three ELEMENTS present in all carbohydrates?

8. What two chemicals are used to test for carbohydrates?

- 9. What color is Benedict's solution?
- 10. What color is iodine?
- 11. What color does Benedict's solution become when it is added to a monosaccharide and heated?
- 12. Does iodine change color when you add it to a monosaccharide? [yes] [no]
- 13. Does Benedict's solution change color when it is heated with a disaccharide? [yes] [no]
- 14. Does iodine change color when it is added to a disaccharide? [yes] [no]
- 15. Does Benedict's change color when it is heated with a poly-saccharide? [yes] [no]
- 16. What color does iodine become when it is mixed with a poly-saccharide?
- 17. What substance would you predict as the result if neither test (Benedict's or Iodine) showed a positive result for saccharides (a "double negative result")?

#### Materials:

- $\boxdot$  test tubes
- ☑ sharpie
- ☑ Iodine solution
- ✓ Hot water bath
- $\square$  monosaccharide solution

- ☑ disaccharide solution
- □
   polysaccharide solution
   □
   apple jule

   □
   polysaccharide solution
   □
   oat solution

   □
   powdered sugar solution
   □
   honey solution
- ☑ Benedict's solution
- ☑ Table Sugar solution
- ☑ apple juice
- ✓ test tube rack

#### **Procedures:**

## **Benedict's Test**

1. Number three clean test tubes 1, 2, and 3. Using a clean dropper for each tube, add the following:

	Tube 1	Tube 2	Tube 3
	<b>20 drops</b> of	<b>20 drops</b> of	20 drops of
	monosaccharide solution	disaccharide solution	polysaccharide
IIIIII			solution
VV	<b>20 drops</b> of	<b>20 drops</b> of	<b>20 drops</b> of
	<b>Benedict's</b> solution	<b>Benedict's</b> solution	<b>Benedict's</b> solution

- 2. Place the three test tubes into the hot water bath for approximately (5) five minutes.
- 3. Remove the tubes from the hot water bath. CAUTION: WATER AND TEST TUBES ARE VERY HOT.
- 4. Observe the **color changes** in the solution.
- 5. Record the color of the solutions in the tubes in the column marked "Benedict's Color after Heating".

#### 6. KEY TO IDENTIFICATION

Color	STAYS	blue/green	<b>green</b>	<b>yellow</b>	orange/red
	blue	(trace)	(little sugar)	(some sugar)	(much sugar)
	-	+	++	+++	++++
Type of saccharide	di- or poly- or NO sugar	mono	mono	mono	mono

### **Iodine Test**

1. Number three clean test tubes 1, 2, and 3. Using a clean dropper for each tube, add the following:

	Tube 1	Tube 2	Tube 3
	20 drops of monosaccharide	20 drops of disaccharide	20 drops of polysaccharide
	solution	solution	solution
<b>MARK</b>	<b>4 drops</b> of	<b>4 drops</b> of	<b>4 drops</b> of
	<b>Iodine</b> solution	Iodine solution	Iodine solution

- 2. Mix the contents of each tube by gently swirling.
- 3. Observe the **color changes** in the solutions.
- 4. Record the color of the solutions in the tubes in the column marked "Iodine Color".

#### DATA TABLE:

Results of Tests with Known Carbohydrates				
Test Tube Number	Carbohydrate Type in solution	Color after heating Benedict's Reagent	Color after Iodine Test	
1	Monosaccharide			
2	disaccharide			
3	polysaccharide			

## **CHEMICAL TESTS – CARBOHYDRATES**

Having tested the carbohydrates and using the results from your tests, you are now ready to test some known carbohydrates. With your results you will be able to classify your carbohydrates as mono- di- or polysaccharides .

#### Procedure:

#### **BENDICT'S TEST**

- 1. Number five (5) clean test tubes: 1,2,3,4,5.
- 2. Using a clean dropper add the following to each test tube:

	TIME T	THE REAL PROPERTY OF THE PROPERTY OF THE REAL PROPE		A REAL PROPERTY AND A REAL
Tube 1	Tube 2	Tube 3	Tube 4	Tube 5
20 drops of	<b>20 drops</b> of	<b>20 drops</b> of <u>table sugar</u> solution	<b>20 drops</b> of	<b>20 drops</b> of
<u>honey</u>	<b>liquid oats</b>		<u>apple juice</u>	<b>powdered sugar</b>
solution	solution		solution	solution
<u>20</u> drops of	<u>20</u> drops of	<u>20</u> drops of	<u>20</u> drops of	<u>20 drops</u> of
Benedict's	Benedict's	Benedict's	Benedict's	<u>Benedict's</u>

- 3. Place the five test tubes into the **hot water bath** for approximately **five (5) minutes**.
- 4. Remove the tubes from the hot water bath. CAUTION: WATER AND TEST TUBES ARE VERY HOT.
- 5. Observe the **color changes** in the solution.
- 6. Record the color of the solutions in the tubes in the column marked "Benedict's Color after Heating".

#### **IODINE TEST**

- 7. Number five (5) clean test tubes: 1,2,3,4,5.
- 8. Using a clean dropper add the following to each test tube:



Tube 1	Tube 2	Tube 3	Tube 4	Tube 5
<b>20 drops</b> of	<b>20 drops</b> of	<b>20 drops</b> of	<b>20 drops</b> of	<b>20 drops</b> of
honey	liquid oats	table sugar	<u>apple juice</u>	powdered sugar
solution	solution	solution	solution	solution
<b>4 <i>drops</i></b> of	<b>4 <i>drops</i></b> of	<b>4 <i>drops</i></b> of	<b>4 <i>drops</i></b> of	<b>4</b> drops of
Iodine	Iodine	Iodine	Iodine	Iodine

- 9. Mix the contents of each tube by gently swirling.
- 10. Observe the **color changes** in the solutions.
- 11. Record the color of the solutions in the tubes in the column marked "Iodine Color".
- 12. Use your results to correctly classify each sugar solution as **a mono-, di-, or polysaccharide.**

Results of Tests with Carbohydrates				
Test Tube Number	Carbohydrate Type in solution	Color after heating Benedict's Reagent	Color after Iodine Test	Identified as monosaccharide disaccharide polysaccharide
1	Honey			
2	Oats			
3	Table sugar			
4	Apple Juice			
5	Powdered Sugar			

#### Conclusion:

You are given an unknown substance. You know it is a saccharide. Describe the test(s) that you would run to determine if it is a <u>monosaccharide</u>, <u>disaccharide</u> or a <u>polysaccharide</u>. Use your own words and be specific in the steps you would take to perform each test. **Look back at your lab for help. BE SPECIFIC!!!!!!!!**